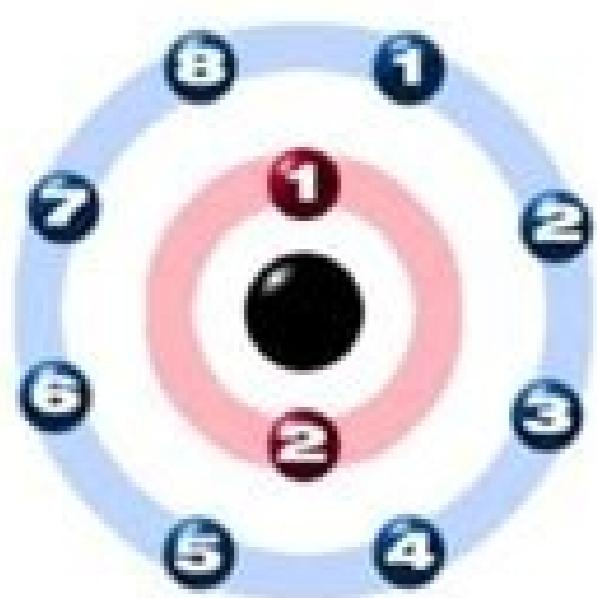


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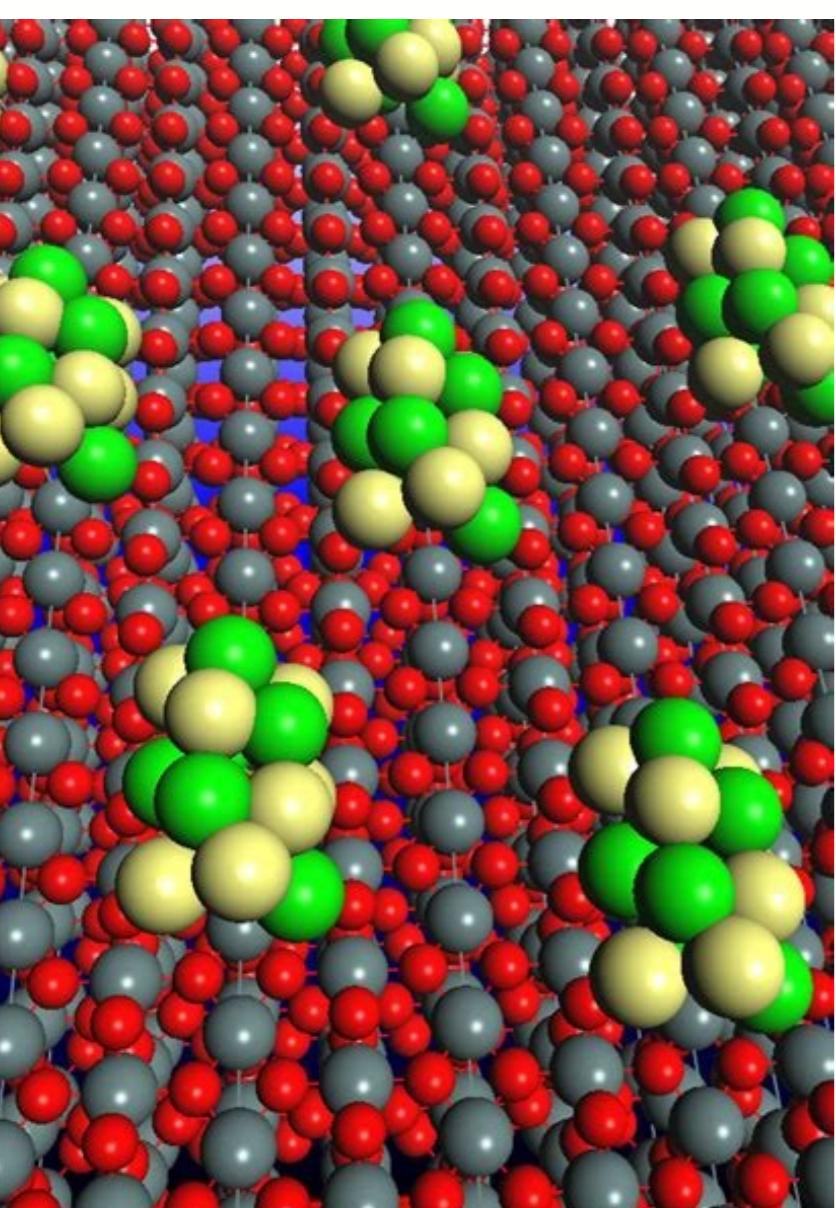
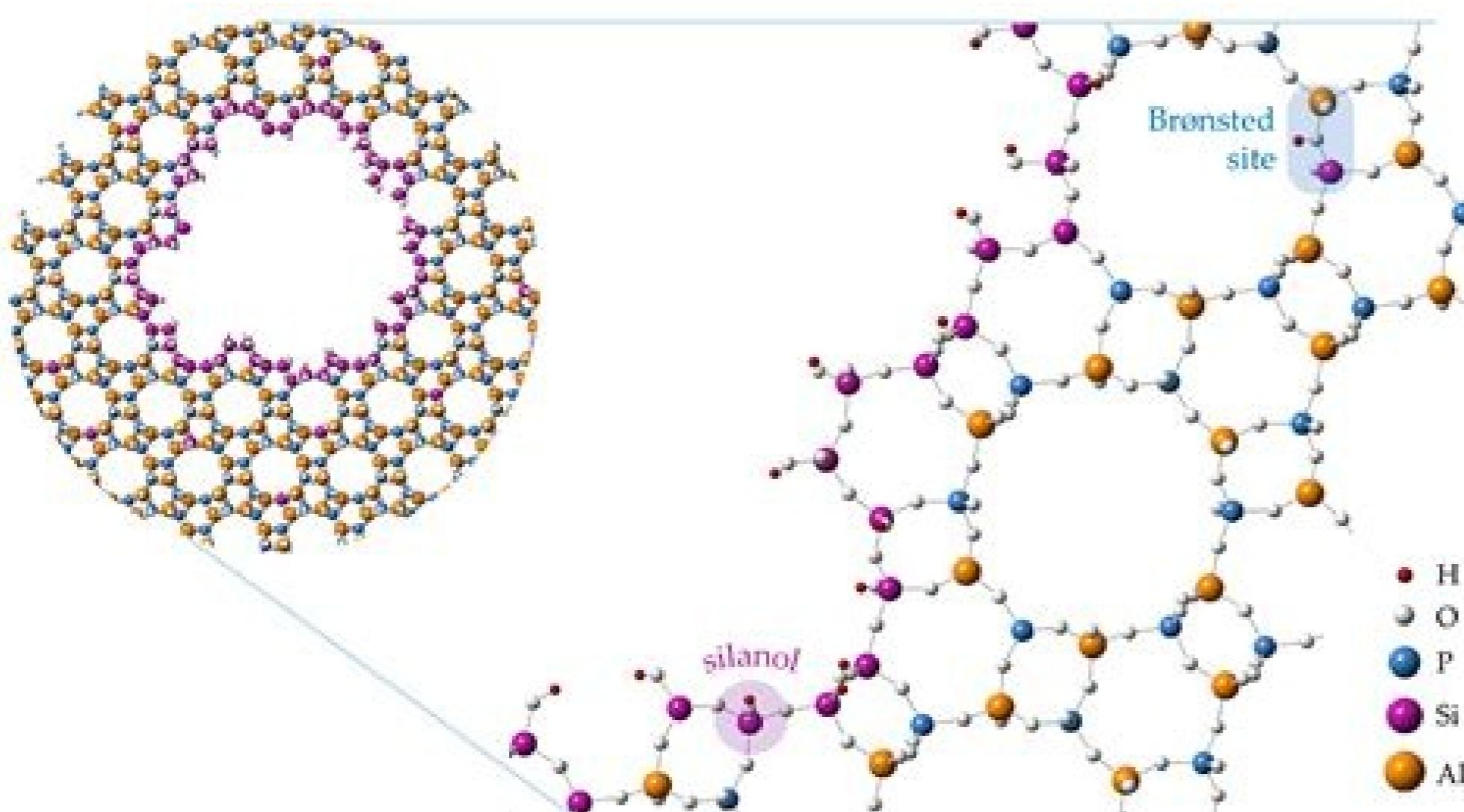
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Write your answers on your handout.

Neon
2 Shells
8 Valence electrons

**Name the element.
Number of shells?
Valence electrons?**



Compare the functions of homogeneous and heterogeneous catalysts. (a) Chemical Mexican Mario Molina (1943-) shared the Nobel Prize in Chemistry in 1995 for his research on (b) the hole of Antarctic ozone. Fortunately, most of the atmospheric chlorine exists like The catalytically inactive shapes Cl_2 and ClO_2 . Since it receives its portion of the Nobel Prize, Molina continued his work in atmospheric chemistry to the mit. Any one of these steps could be slow and therefore can serve as the rate of determination of the rate . It is not permanently exhausted; then, it acts as a catalyst. (C) Atoms spread through the surface and form new ties H when they collide. A catalytic converter allows the combustion of all the compounds containing carbon carbon, while at the same time they reduce the production of nitrogen oxide and other pollutants in the emissions of petrol engines. One of these reactions is catalytic hydrogenation, the process by means of Which hydrogen is added through a link Alkene $\text{C} = \text{C}$ to afford the saturated alkane product. The end of chapter chemistry exercises represent the increase in the reaction rate caused by a catalyst. Explain your answer. Consider this scenario and answer the following questions: chlorine atoms deriving from the decomposition of chlorofluorometan, such as CCl_2F_2 , catalyzes the decomposition of ozone into the atmosphere. Modern three-way catalytic converters possess a surface impregnated by a platinum-rhodium catalyst, which catalyzes the conversion nitric oxide into dundricene and oxygen and the conversion of carbon monoxide and hydrocarbons as the octane into carbon dioxide and water vapor: [math display="block">\begin{array}{r} \text{r} @ \{\} \longright\} \text{l} \\ \text{2 text \{no\}}_2(\text{g}) & \text{& text \{n\}}_2(\text{g}); + 2 \text{ text \{or\}}_2[0.5em] 2 \text{ text \{co\}}(\text{g}); + \text{ text \{o\}}_2(\text{g}) \text{ and } 2 \text{ text \{co\}}_2(\text{g}) [0.5em] 2 \text{ text \{c\}}_8 \text{ text \{h\}}_{18}(\text{g}); + 25 \text{ text \{o\}}_2(\text{g}) \text{ and } 16 \text{ text \{co\}}_2(\text{g}); + 18 \text{ text \{h\}}_2(\text{g}) \text{ end \{array\}} / \text{ in latex} \end{array} to be as The most as possible, catalytic converters are preheated by an electric heater. As shown, the catalyst path provides for a two-phase mechanism (known as the presence of two transition states) and an intermediate species (represented by the valley between the two transitions of the states). Other significant industrial processes involving the use of heterogeneous catalysts include the preparation of sulfuric acid, ammoniac preparation, ammoniac oxidation

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